

# Physics Crib Sheet: Topic 1

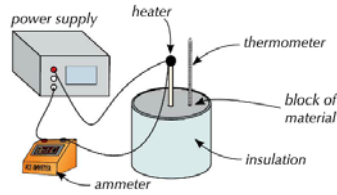
**A** Some energy resources can be replenished as they are used (**renewable**) but others can not (**non renewable**). The main uses of energy resources are **transport** and **heating**. The demand for energy is increasing and scientists are looking for alternative 'greener' and renewable resources.

**C** Specific heat capacity is the amount of energy needed to raise the temperature of 1 kg of a substance by 1 °C.

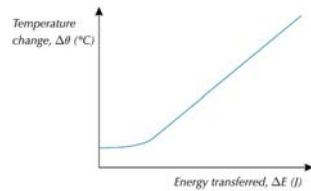
$$\Delta E = mc\Delta\theta$$

$\Delta E =$  change in thermal energy (J)       $\Delta\theta =$  temperature change (°C)  
 $m =$  mass (kg)       $c =$  specific heat capacity (J/kg°C)

**D** Core practical: Investigating Specific heat capacity



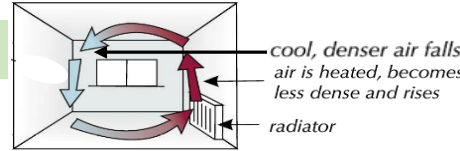
Use the current and voltage reading to calculate **power**. Use this to calculate how much energy has been transferred by the heater. Assuming no energy has been dissipated you can plot a graph:



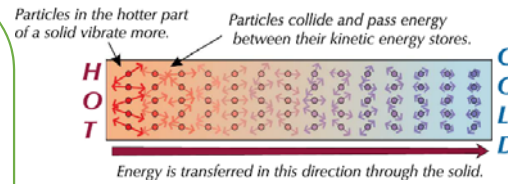
You can find the specific heat capacity of the block using the gradient of the linear part of your graph. The gradient is  $\Delta\theta \div \Delta E$ , so since  $\Delta E = mc\Delta\theta$ , the gradient is  $1 \div mc$ . So the specific heat capacity of the material of the block is:  $1 \div (\text{gradient} \times \text{the mass of the block})$ .

$$\text{efficiency} = \frac{\text{useful output energy transfer}}{\text{total input energy transfer}}$$

**E** Convection is where energetic particles move away from hotter to cooler regions



**C** Conduction is the process by which vibrating particles transfer energy to neighbouring particles.



**F** Thermal conductivity is a measure of how quickly energy is transferred. Thermal insulators can be used to reduce the amount of heat transferred in buildings:



**M** Lubricants reduce friction between moving objects which reduces energy dissipated. In any energy transfer some energy will be dissipated. This is measured by efficiency

Energy store	Objects with energy in this store
Kinetic	Anything moving has energy in its kinetic energy store.
Thermal	Any object. The hotter it is, the more energy it has in this store. You may also see thermal energy stores called internal energy stores.
Chemical	Anything that can release energy by a chemical reaction, e.g. food, fuels.
Gravitational Potential	Anything that has mass and is inside a gravitational field.
Elastic Potential	Anything that is stretched (or compressed) e.g. springs.
Electrostatic	Anything with electric charge that is interacting with another electric charge — e.g. two charges that attract or repel each other.
Magnetic	Anything magnetic that is interacting with another magnet — e.g. two magnets that attract or repel each other.
Nuclear	Atomic nuclei have energy in this store that can be released in nuclear reactions.

**H**  $P = \text{power (W)}$        $P = \frac{E}{t}$        $P = \text{power (W)}$        $P = \frac{W}{t}$

$E =$  energy transferred (J)       $W =$  work done (J)  
 $t =$  time (s)       $t =$  time (s)

**J** Work done is energy transferred from one store to another. The principle of **conservation of energy** states that energy can be transferred usefully, stored, dissipated but never created or destroyed. **Power** is the rate of energy transferred.

**K**  $\text{efficiency} = \frac{\text{useful power output}}{\text{total power input}}$

**L**  $E_k = \text{kinetic energy (J)}$        $E_k = \frac{1}{2}mv^2$        $v =$  speed (m/s)  
 $m =$  mass (kg)

**N**  $E_p = \text{gravitational potential energy (J)}$        $E_p = mgh$        $g =$  gravitational field strength (N/kg)  
 $h =$  height (m)  
 $m =$  mass (kg)

**O** Energy transferred to falling objects  
 When something falls, energy from its gravitational potential energy store is transferred to its kinetic energy store. The further it falls, the faster it goes. For a falling object where there's no air resistance, you can use the principle of conservation of energy to get:  
 Energy lost from the object's store = Energy gained in the kinetic energy store